## CHEMISTRY 163A FALL 2002

## PROBLEM SET

1. A simple differential equation which is important to quantum mechanics is

$$\frac{\mathrm{df}(t)}{\mathrm{dt}} = \frac{-i}{\hbar} E f(t)$$

where E,  $\hbar$ , and i [i = (-1)<sup>1/2</sup>] are constants. Solve this differential equation for f(t) with the boundary condition f(0) = 1.

- 2. In this problem we demonstrate that classical mechanics and classical electrodynamics would predict the rapid collapse of the hydrogen atom in the Rutherford model.
  - (a) From classical electrodynamics, the intensity of radiation given off by an electron circulating within an angular velocity  $\omega$  at a distance r from the nucleus us:

$$I = \frac{dE_{rad}}{dt} = -\frac{dE_{H}}{dt} = \frac{e^{2}\omega^{4}r^{2}}{6\pi\epsilon_{0}c^{3}}$$

where  $E_{rad}$  is the energy of the emitted radiation and  $E_{H}$  is the energy of the hydrogen

atom. Use the chain rule to find an expression for  $\frac{dr}{dt}$  in terms of  $\frac{dE_{rad}}{dt}$  and  $\frac{dr}{dE_{H}}$ .

(b) Using the result from (a) and relationships derived in class for the Rutherford hydrogen

atom, show that 
$$\frac{dr}{dt} = \frac{-4}{3Zc^3} \left( \frac{2Ze^2}{8\pi\epsilon_0 rm} \right)^2$$

- (c) Solving this differential equation, calculate the time it would take for an electron initially at the Bohr radius,  $a_0 = 5.29 \times 10^{-11}$  m to collapse into the nucleus (you should look up a reasonable estimate for the radius of the nucleus).
- 3. Problem #1- 23 McQ (parts a and b) and
  - c) Calculate the De Broglie wavelength of an Edsel weighing 900 kg (~ 1 ton) and traveling 0.027 km/sec (~60 mph). What is the implication of the relative magnitudes of the wavelengths calculated in parts a, b, and c?
- 4. In lecture we stated that the relativistic energy of a free electron was

$$E = (m_0^2 c^4 + p^2 c^2)^{1/2}$$

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where

$$p = \frac{m_0 v}{(1 - v^2 / c^2)^{1/2}}$$

is the relativistic expression for momentum.

Show that in the correspondence limit,  $v \ll c$ , the above expression becomes

$$E = m_0 c^2 + 1/2 m_0 v^2$$

- b) What is the significance of each of the terms in the above expression?
- 5. A mass (m) attached to a spring is one, idealized, example of a harmonic oscillator. Here the mass is acted upon by the form F = -kx, where x = 0 is the resting (equilibrium) position of the spring and k and the spring constant.
  - a) Using Newton's law, F = ma, write a differential equation for the motion of a classical harmonic oscillator.
  - b) If the spring is fully extended to a length A at time t = 0, what is

$$\left(\frac{\mathrm{dx}}{\mathrm{dt}}\right)_{\mathrm{t}=0} = ?$$

- Show that  $x(t) = A\cos\left[\left(\frac{k}{m}\right)^{\frac{1}{2}} t\right]$  satisfies the equation of motion and the boundary conditions in part (b).
- d) What is the length  $\tau$  of one period of vibration? Will a larger spring constant increase or decrease the period of oscillation? What effect will increasing the mass have on the period of vibration?
- e) What is the average kinetic energy:

(i.e., evaluate) 
$$\overline{K.E.} = \frac{1}{\tau} \int_0^{\tau} \frac{1}{2} m[v(t)]^2 dt$$

- \*f) What is the average potential energy, assuming U(x=0) = 0 (you set up that integral and evaluate)?
- \*g) If you've gotten part (e) correct, there is a simple relationship between K.E. and P.E. for the harmonic oscillator. How could this have been obtained without direct evaluation of the integrals in part (e)?