PROBLEMS

CHEMISTRY 163A

- 14. McQ #3-16
- 15. McQ #3-17;

Does σ_x^2 increase or decrease as *n* gets larger?

From "looking" at the dependence of PIB wave functions on *n*, comment on the behavior of σ_x^2 as $n \to \infty$.

- 16. McQ #3-26
- 17. Translational motions of "real" atoms:

A 1 liter cubic box contains He (g) at 300°K. To calculate the translational quantum states of the atoms of He(g) treat their potential energy as being zero inside the box and ∞ outside the container.

- a. What is the <u>average</u> kinetic energy of a helium atom at this temperature (from thermodynamics class or be nice to your TA)?
- b. What value of the quantum number *n* corresponds to the quantum state of a He atom with the above energy? [After choosing the appropriate box length, assume a state for which $n_x = n_y = n_z = n$.]
- c. What is the ratio of the transition energy from state $n \rightarrow n+1$ to the translational energy of the atom for the "average energy" He atom at 300°K?

i.e.
$$\frac{\Delta E_{n \to n+1}}{E_n}$$

(for E_{n+1} , $n_x = n+1$, $n_y = n$, $n_z = n$)

d. What is the "bottom line" which this result illustrates vis à vis quantum mechanics, classical mechanics, and translational energy?

(bottom line)