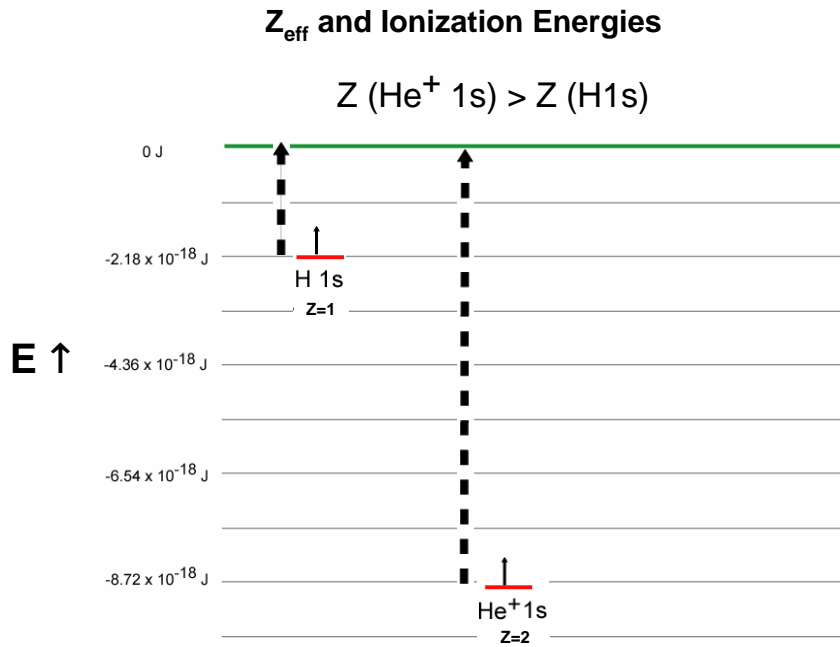
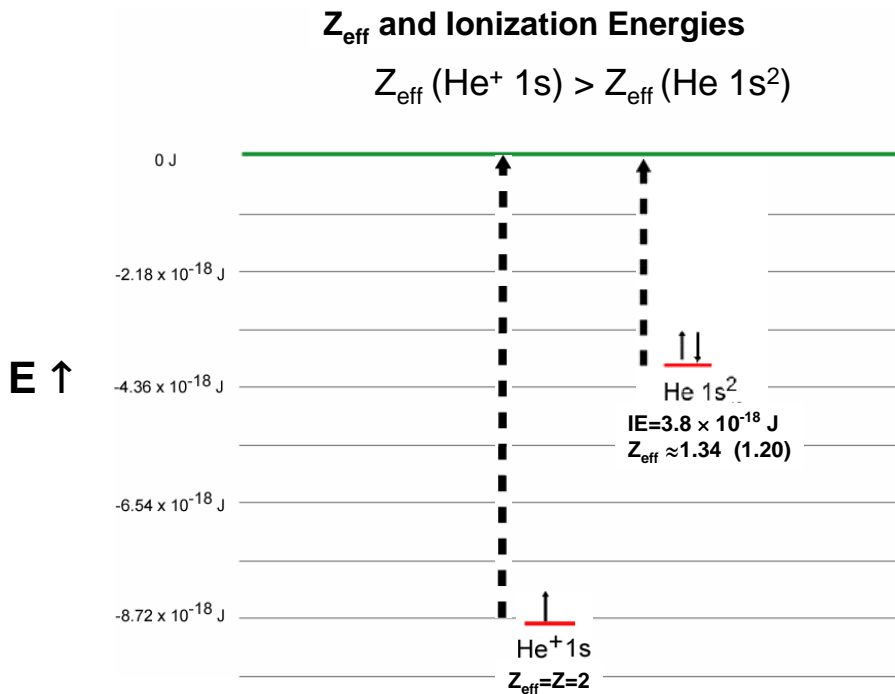


Energy of $H\ 1s$ vs $He^+\ 1s$ (HO 12.1)



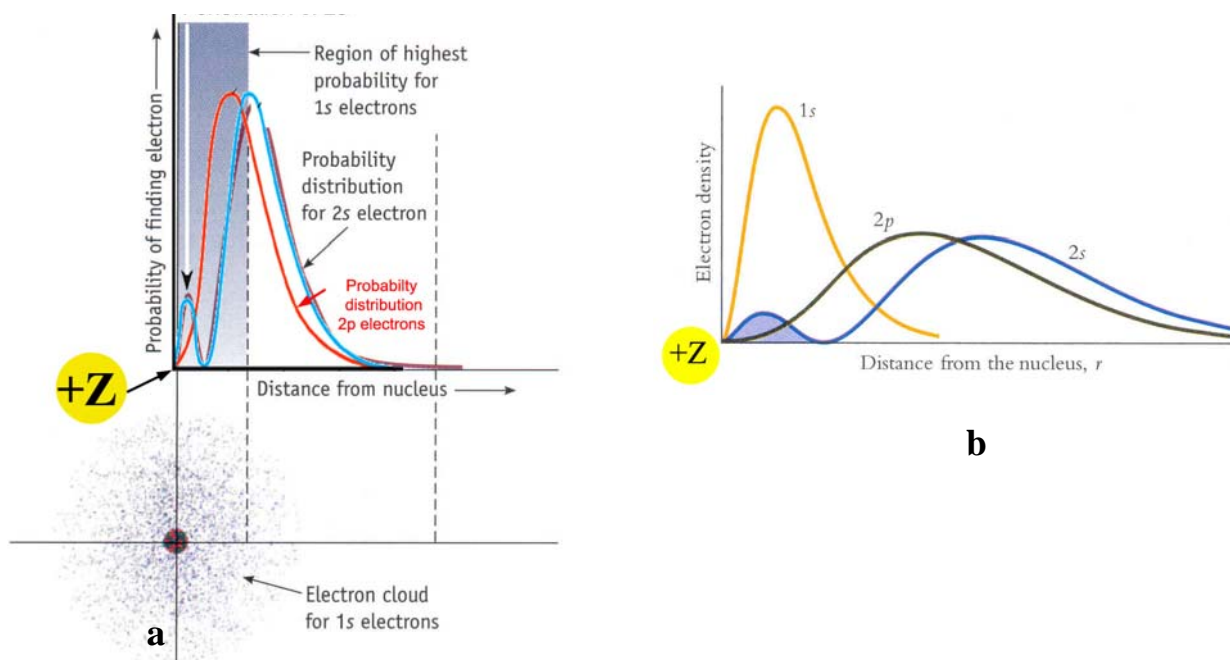
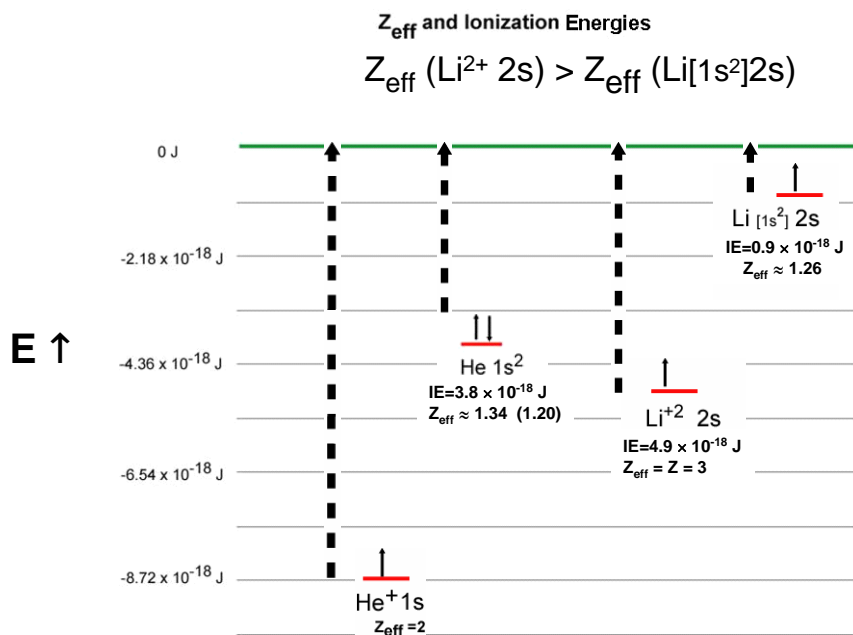
HO Figure 12.1

Energy of $He\ 1s^2$ vs $He^+\ 1s$ (HO fig 12.2)



HO Figure 12.2

Energy of $\text{Li}^{2+} 2s$ vs $\text{Li} [1s^2] 2s$ (HO Fig. 12.3)

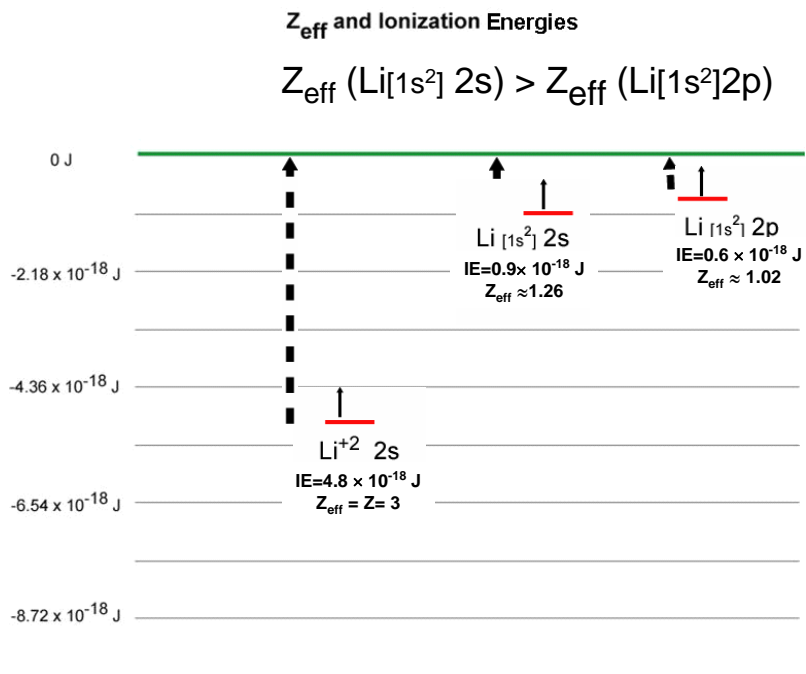


a. adapted from **Chemistry & Chemical Activity 5th ed**, by Kotz and Treichel, Thompson Brooks/Cole (2003)

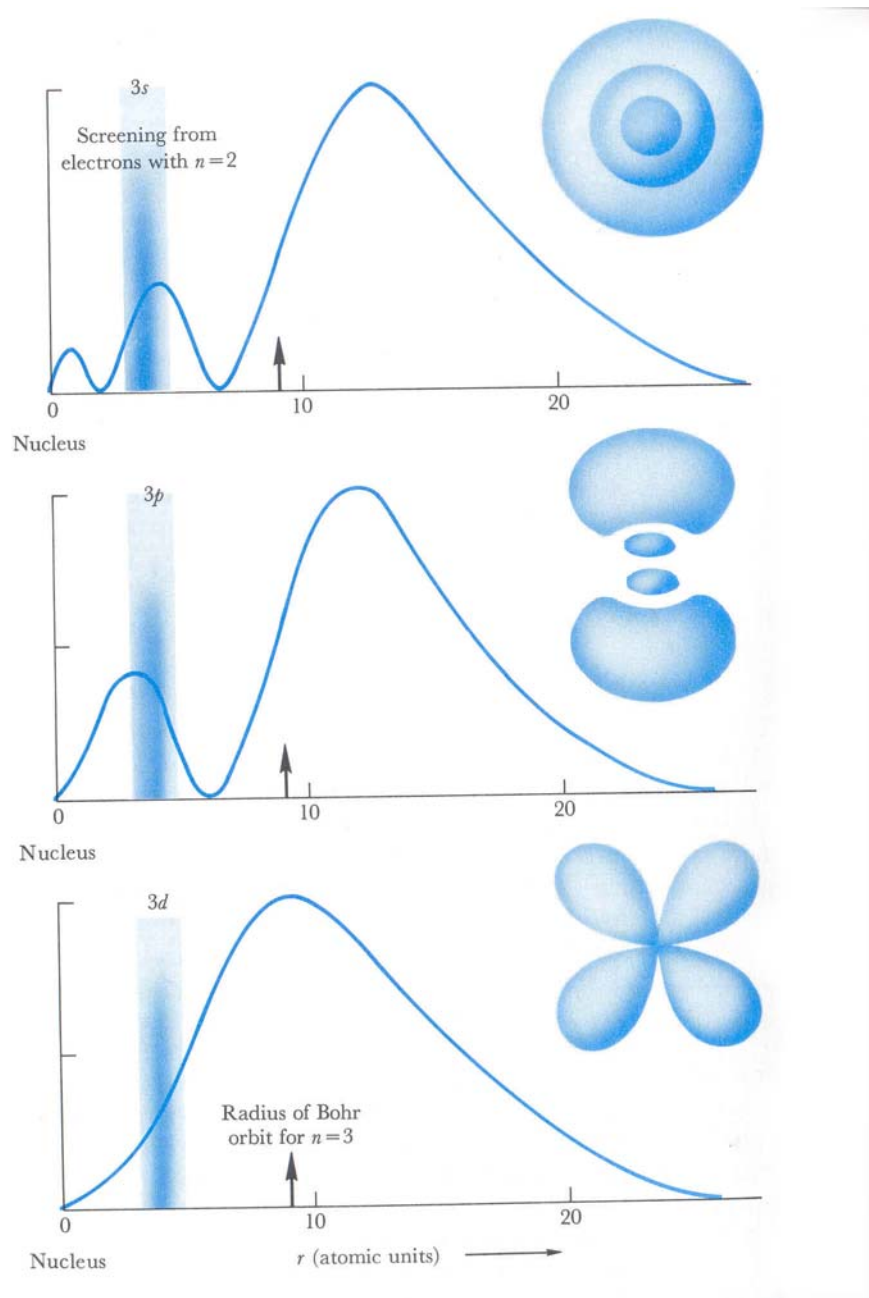
and **Chemistry: The Molecular nature of Matter and Change, 3rd ed**, Silberg, McGraw Hill, (2003)

b. from **Chemistry 3rd ed**, by Olmsted and Williams, Wiley (2002).

Energy of Li $[1s^2] 2s$ vs Li $[1s^2] 2p$ (HO Fig. 12.5): penetration



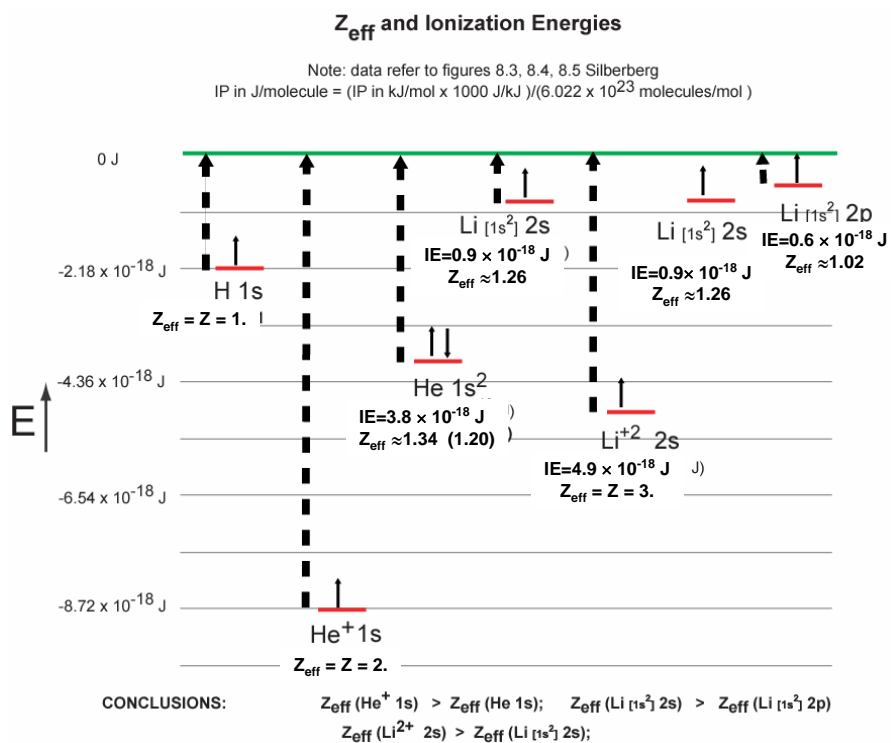
HO Figure 12.5



from **Chemical Principles**, Dickerson, Gray, Haight, Benjamin Cummings, 1979.

HO Figure 12.6

Z_{eff} and ionization potentials



HO Figure 12.7