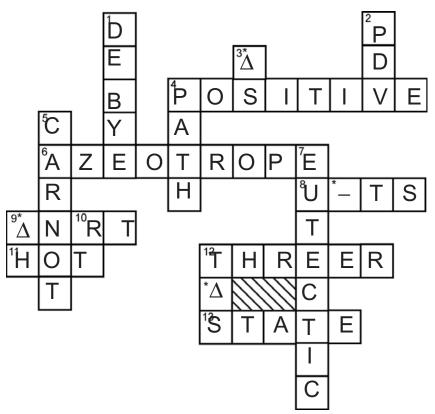
Thermodynamic crossword

Fill in this thermodynamic crossword. The boxes marked with * are mathematical symbols; all other entries are letters.



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| ACROSS | | Down | |
|--------|---|------|--|
| 4. | If a reaction is endothermic ΔH is· | 1. | One name associated with the theory for calculation of activity coefficients of electrolytes |
| 6. | Constant boiling liquid solution | 2. | Work of expansion done on surroundings |
| 8. | (Helmholtz free energy) A ≡ | 3. | Change in order-disorder |
| 9. | For isothermal reaction involving only ideal gasses ΔH= ΔU + | 4. | $\int\limits_{initial}^{final}dG$ is independent of |
| 11. | In an adiabatic compression an ideal gas will become | 5. | Famous French cyclist and efficiency expert |
| 12. | One mole of monatomic ideal gas is heated from 100K to 738.9056099K at constant volume. What is ΔS? | 7. | Constant melting solid mixture |
| 13. | If dA is an exact differential then A is a function | 9 | q at constant pressure |
| | | 10. | For any gas $\lim_{\frac{n}{p}\to 0} \left(\frac{pV}{n}\right)$ |
| | | 12. | q,,, for constant T |